



**CÂMARA DOS DEPUTADOS**

**COMISSÃO DE MEIO AMBIENTE E DESENVOLVIMENTO SUSTENTÁVEL**

**OF-CONV N.º 502**

# **“Gás de Xisto”: Desafios e Perspectivas**



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**Instituto de Química, UFBA;  
INCT – Energia e Ambiente**

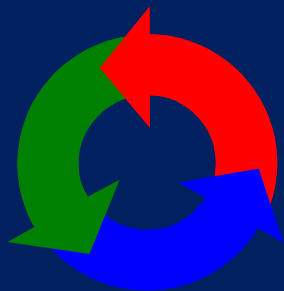
**([jailsondeandrade@gmail.com](mailto:jailsondeandrade@gmail.com))**



**Brasília, DF, 05 de dezembro de 2013**

**Energia**

**Alimentos**



**Ambiente**

**Pobreza**

**Água**

**Popul**

# Sustentabilidade

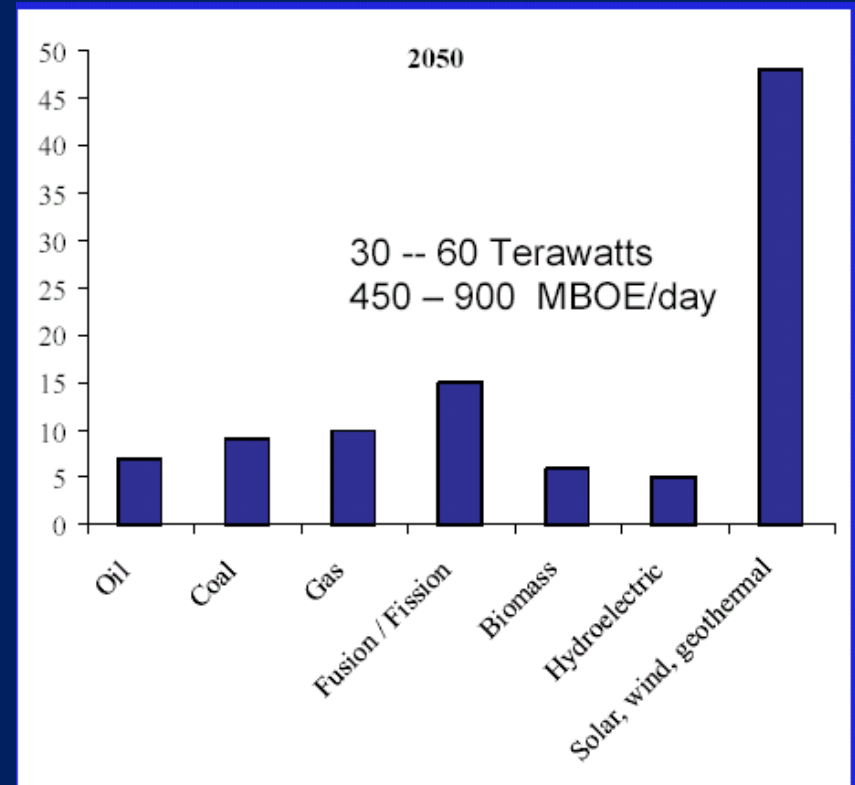
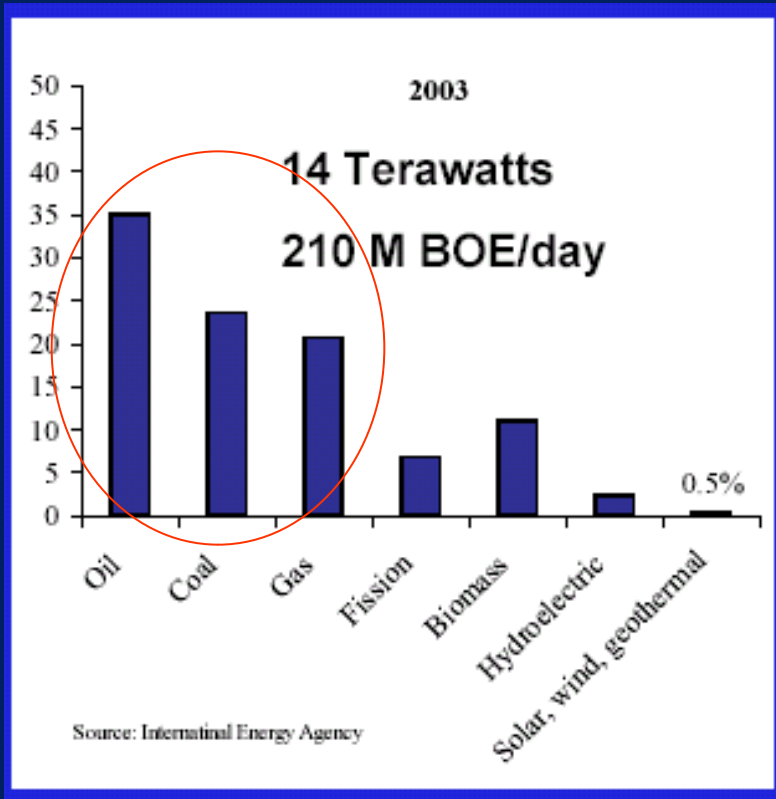


**Educação**

**Democracia**

**Terrorismo, Guerra, Violência**

# O Desafio Energético



A Base da Prosperidade  
Século XX = Óleo  
Século XXI = ???



que significa muito mais do que proteger refinarias e oleodutos contra ataques terroristas.

## **Segurança energética**

pode ser melhor compreendido como a capacidade de manter a máquina global funcionando, isto é, produzindo combustíveis e eletricidade suficientes, a preços acessíveis, para que todos os países possam, pelo menos, manter sua economia operando e o seu povo alimentado.



# Segurança energética

**Sustentabilidade** (quanto tempo o petróleo do mundo, gás natural, urânio e fornecimento de carvão durarão);

**Globalização** (o impacto do rápido crescimento da demanda de energia nos países em desenvolvimento e as respectivas implicações econômicas, políticas e ambientais);

**Regulamentação**

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# Educação

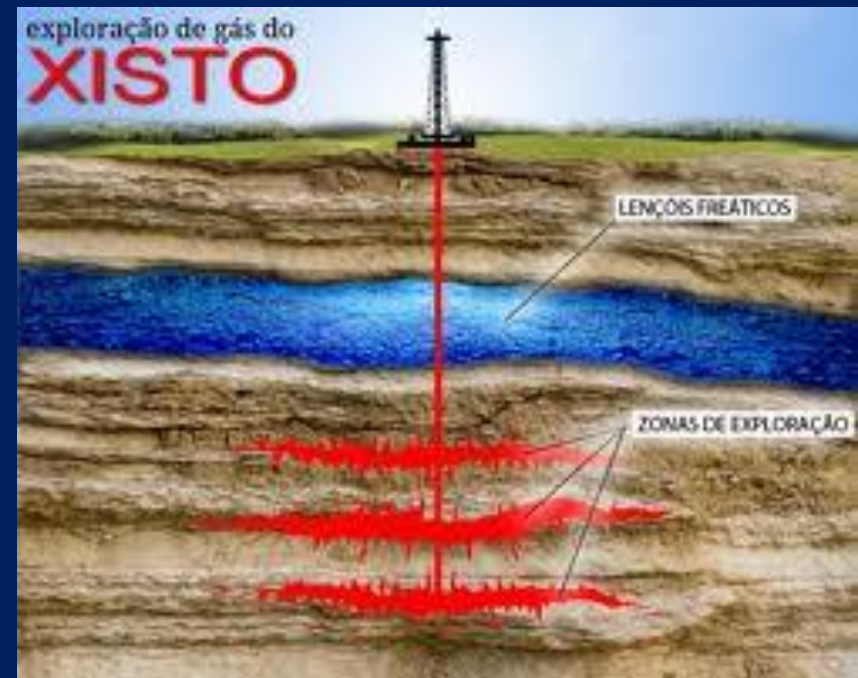
questões energéticas na qualidade de vida, ou o equilíbrio percebido entre consumo e responsabilidade ambiental), e o

**Potencial de desenvolvimento tecnológico** (o equilíbrio entre as tecnologias que já estão disponíveis, aqueles que estarão disponíveis em um futuro próximo, e aqueles que podem tornar-se em longo prazo em "mudanças do jogo").

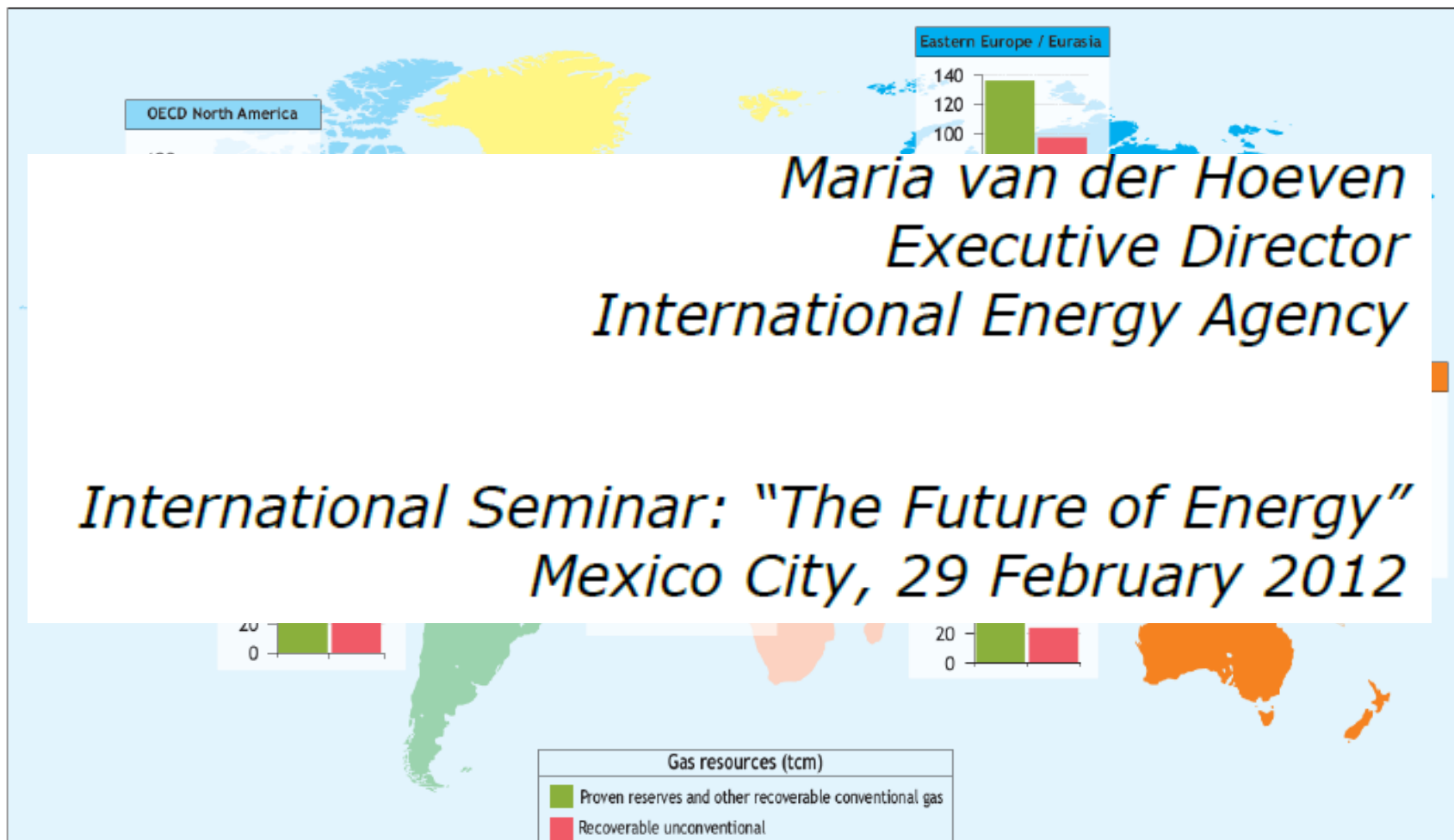


# “Shale Gas” / Gás de xisto

É o gás natural o que pode ser encontrado preso dentro de formações de XISTO, nome genérico de vários tipo de rochas de folhelo.



# Are we entering a Golden Age of Gas ?



***Natural gas can enhance security of supply: global resources exceed 250 years of current production; while in each region, resources exceed 75 years of current consumption***

# BACIAS SEDIMENTARES COM POTENCIAL DE GÁS NÃO CONVENCIONAL NO BRASIL

Global

Top reserv



U.S. 24.4

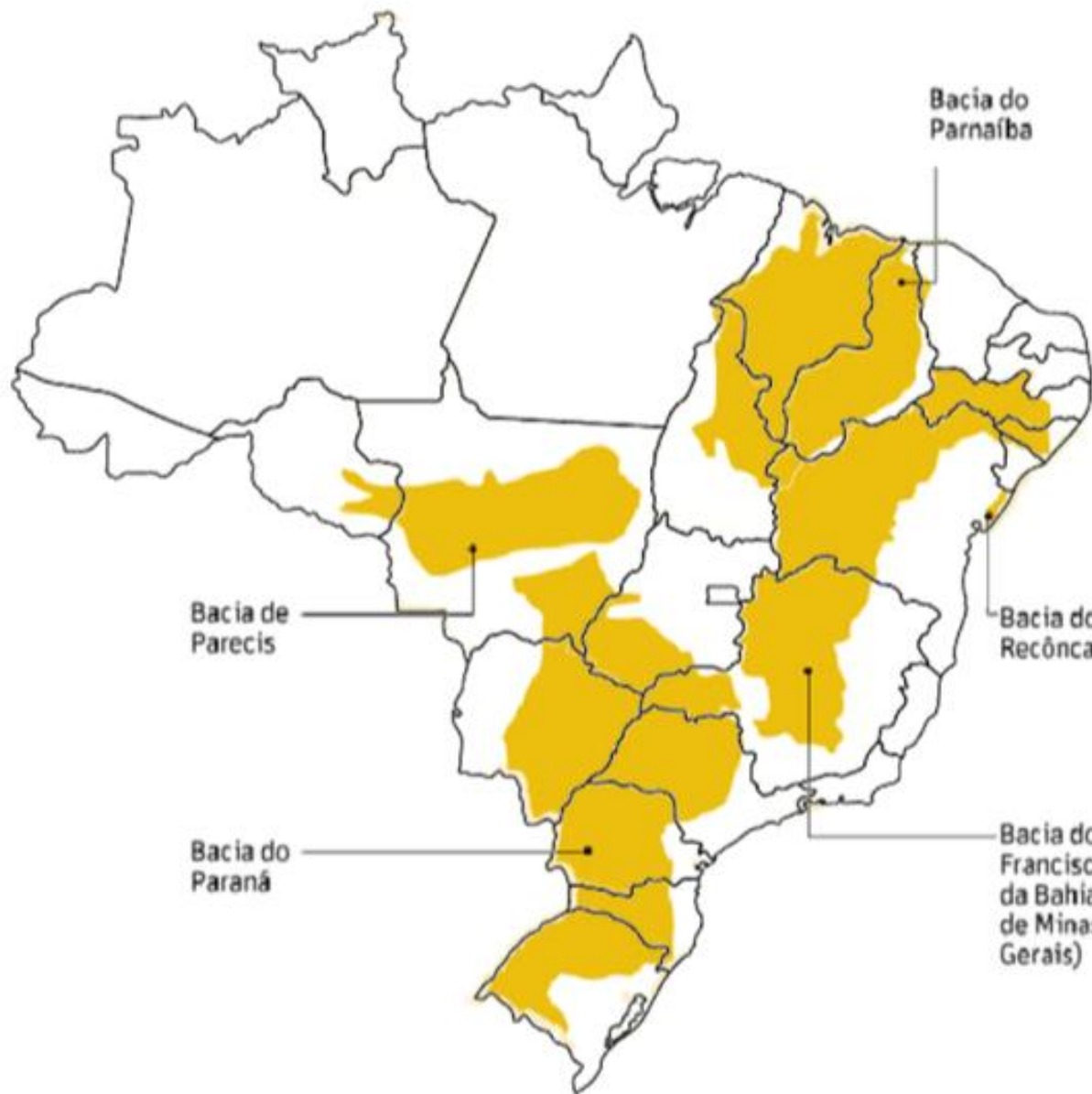
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Source: EIA base

Reuters graphic/C



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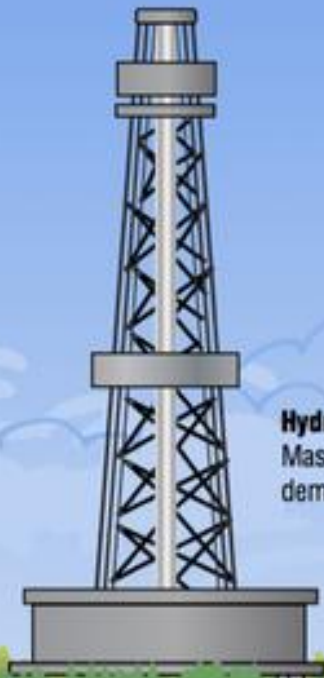


REUTERS

\*Considera apenas avaliação da bacia do Paraná Fontes: CBIE(Centro Brasileiro de Infraestrutura)/AIE(Agência Internacional de Energia)/ANP (Agência Nacional do Petróleo, Gás Natural e Biocombustíveis)



# Where Shale Gas Comes From



**Hydraulic fracturing in shale:**  
Massive hydraulic fracturing (MHF)  
demonstrated by DOE in 1977.

**Unconventional natural gas:**  
Pre-commercial resource incentivized by 1980-2002 production tax credit.

**Diamond-studded drill bits:**  
Partnership between General Electric and  
the Energy Research and Development  
Administration, precursor to DOE.

**Directional and horizontal drilling:** Early directional shale drilling  
patented by federal Morgantown Energy Research Center engineers in 1976.

**Microseismic  
imaging and  
electromagnetic  
telemetry:**  
Developed by  
Sandia National  
Laboratories for  
non-shale applications.

**Multi-fracture  
Horizontal Drilling:**  
First commercial  
demonstration from  
DOE-private  
venture in 1986.

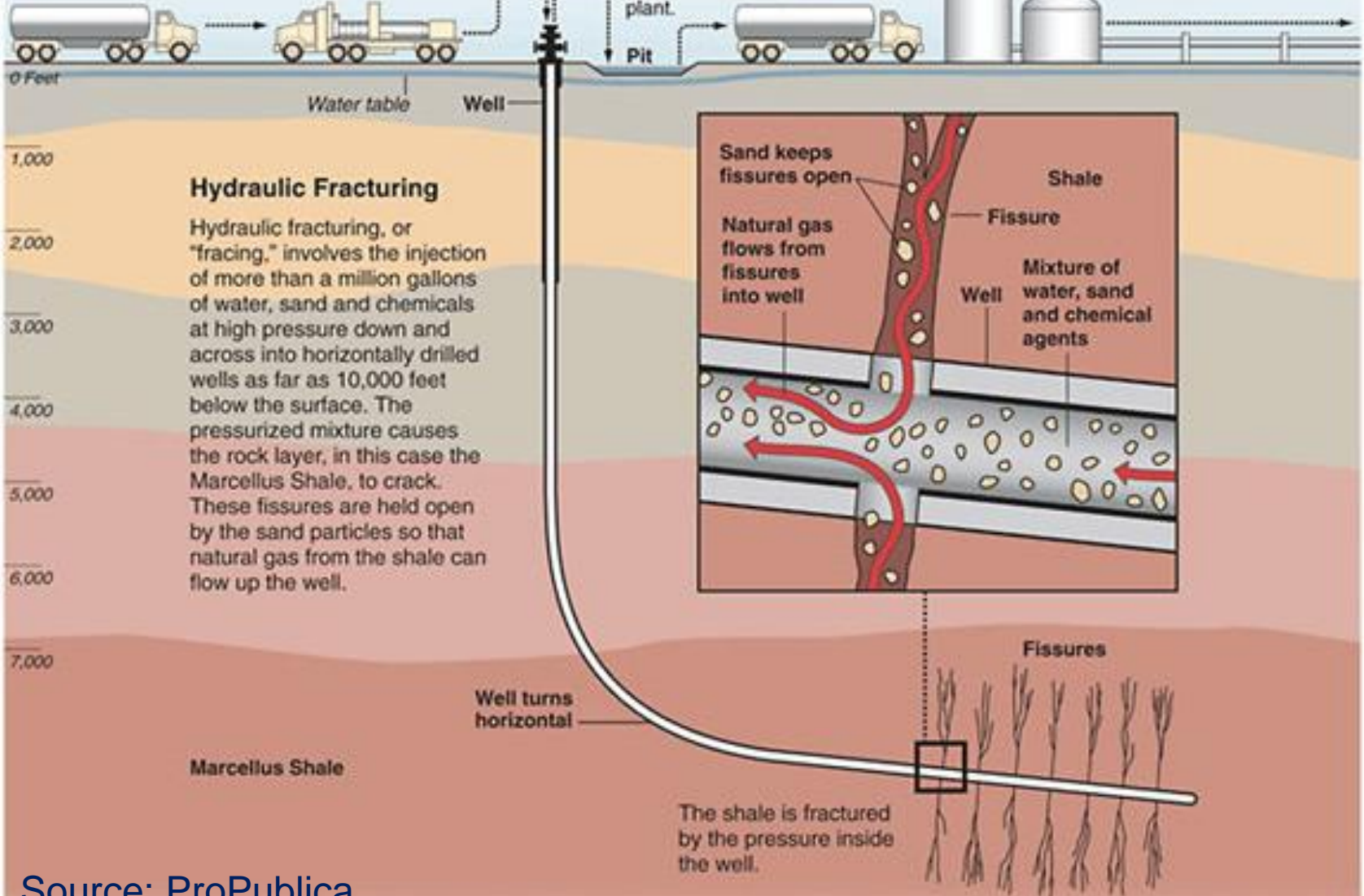
Roughly 200 tanker trucks deliver water for the fracturing process.

A pumper truck injects a mix of sand, water and chemicals into the well.

Natural gas flows out of well.

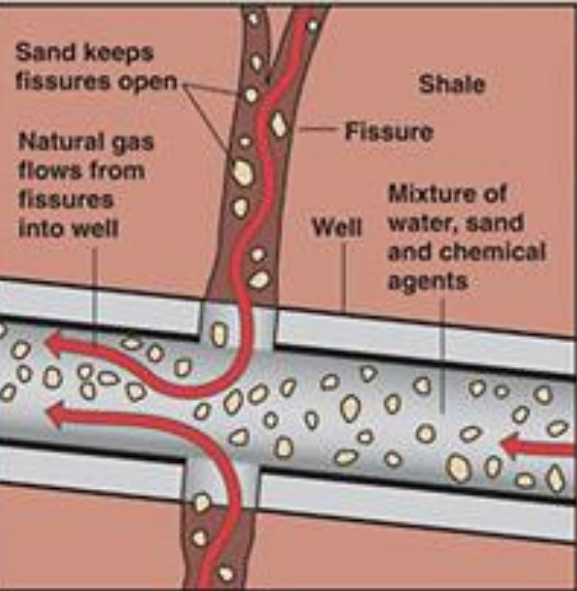
Recovered water is stored in open pits, then taken to a treatment plant.

Storage tanks  
Natural gas is piped to market.



### Hydraulic Fracturing

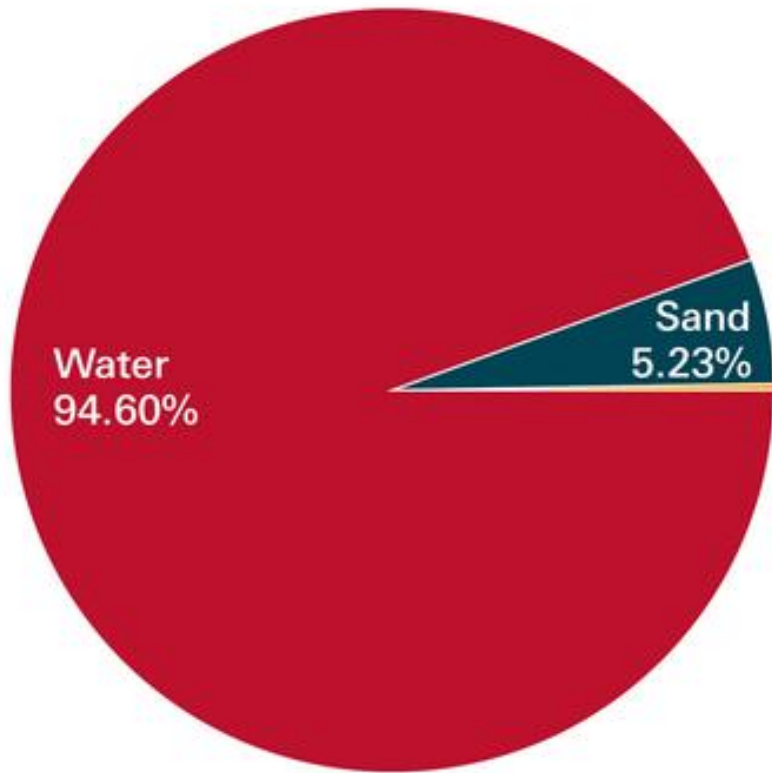
Hydraulic fracturing, or "fracing," involves the injection of more than a million gallons of water, sand and chemicals at high pressure down and across into horizontally drilled wells as far as 10,000 feet below the surface. The pressurized mixture causes the rock layer, in this case the Marcellus Shale, to crack. These fissures are held open by the sand particles so that natural gas from the shale can flow up the well.



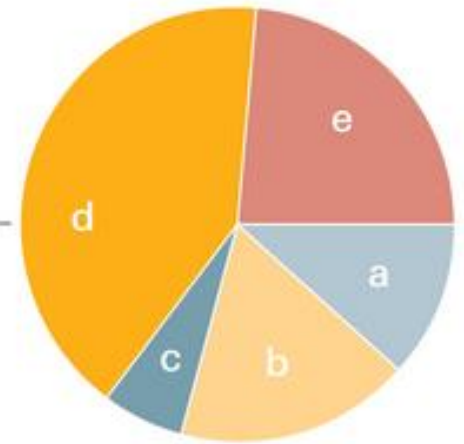
Source: ProPublica,

<http://www.propublica.org/special/hydraulic-fracturing-national>

Graphic by Al Granberg



Additives  
0.17%



- a. Scale inhibitor
- b. Acid
- c. Biocide
- d. Friction reducer
- e. Surfacant

## Chemical additives: Types and volumes

The numbers and volume of additives used may differ from one shale play to another, depending on the properties and the depth of the target shale, among other things. Therefore, a large variety of fracturing fluid additives exists (Tab.1).

Tab. 1: An overview of common additive classes, their purpose and some examples.

Additive Class	Purpose	Examples
Biocide	Avoiding growth of bacteria and other fauna	Terpenes, isothiazolinones (e.g. 1,2-benzisothiazol-3-(2H)-one or 2-methyl-4-isothiazolin-3-one)
Buffer	pH control	Anorganic acids and bases (e.g. hydrofluoric acid, ammonium bisulfite)
Breaker	Reducing viscosity, enhanced fluid retrieval	Sulfates, peroxides (e.g. Ammonium persulfate, calcium peroxide)
Corrosion Inhibitor	Protect casing and equipment	Acids, alcohols, sulfites, (e.g. 2-butoxyethanol, amine bisulfite)
Crosslinker	Support gel formation, increase viscosity for proper downhole transportation of sand.	Borates, transition metals in combination with complexing agents (e.g. zirconiumoxide, -sulfate)
Friction Reducer	Creates laminar instead of turbulent flow	Polyacrylamide, petroleum distillates, e.g. aromatic hydrocarbons (benzene, toluene)
Gelling Agent	Support gel formation, increase viscosity for proper downhole transportation of sand, ideal proppant carriage	Guar gum, hydroxyethylcellulose, polymers (e.g. acrylamidcopolymers, vinylsulfonates)
Scale Inhibitor	Avoid precipitates from mineralic scalings that may build up at the inner wall of the casing or in the wellhead	Acids, phosphonates, (e.g. dodecylbenzene, sulfonic acid, calcium phosphonate)
Surfactant	Emulsification and salinity tolerance	Amines, glycol ethers, nonylphenol ethoxylates

<http://www.shale-gas-information-platform.org/categories/water-protection/the-basics/fracturing-fluids.html>



## Impact of Shale Gas Development on Regional Water Quality

R. D. Vidic,<sup>1\*</sup> S. L. Brantley,<sup>2</sup> J. M. Vandenbossche,<sup>1</sup> D. Yoxtheimer,<sup>2</sup> J. D. Abad<sup>1</sup>

**Background:** Natural gas has recently emerged as a relatively clean energy source that offers the opportunity for a number of regions around the world to reduce their reliance on energy imports. It can also serve as a transition fuel that will allow for the shift from coal to renewable energy resources while helping to reduce the emissions of CO<sub>2</sub>, criteria pollutants, and mercury by the power sector. Horizontal drilling and hydraulic fracturing make the extraction of tightly bound natural gas from shale formations economically feasible. These technologies are not free from environmental risks, however, especially those related to regional water quality, such as gas migration, contaminant transport through induced and natural fractures, wastewater discharge, and accidental spills. The focus of this Review is on the current understanding of these environmental issues.

**Advances:** The most common problem with well construction is a faulty seal that is emplaced to prevent gas migration into shallow groundwater. The incidence rate of seal problems in unconventional gas wells is relatively low (1 to 3%), but there is a substantial controversy whether the methane detected in private groundwater wells in the area where drilling for unconventional gas is ongoing



READ THE FULL ARTICLE ONLINE

<http://dx.doi.org/10.1126/science.1235009>

Cite this article as R. Vidic *et al.*, *Science* 340, 1235009 (2013). DOI: 10.1126/science.1235009

### ARTICLE OUTLINE

Cause of the Shale Gas Development Surge

Methane Migration

How Protective Is the “Well Armor”?

The Source and Fate of Fracturing Fluid

Appropriate Wastewater Management Options

Conclusions

### BACKGROUND READING



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## Groundwater Protection and Unconventional Gas Extraction: The Critical Need for Field-Based Hydrogeological Research

by R.E. Jackson<sup>1</sup>, A.W. Gorody<sup>2</sup>, B. Mayer<sup>3</sup>, J.W. Roy<sup>4</sup>, M.C. Ryan<sup>3</sup>, and D.R. Van Stempvoort<sup>4</sup>

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### Abstract

Unconventional natural gas extraction from tight sandstones, shales, and some coal-beds is typically accomplished by horizontal drilling and hydraulic fracturing that is necessary for economic development of these new hydrocarbon resources. Concerns have been raised regarding the potential for contamination of shallow groundwater by stray gases, formation waters, and fracturing chemicals associated with unconventional gas exploration. A lack of sound scientific hydrogeological field observations and a scarcity of published peer-reviewed articles on the effects of both conventional and unconventional oil and gas activities on shallow groundwater make it difficult to address these issues. Here, we discuss several case studies related to both conventional and unconventional oil and gas activities illustrating how under some circumstances stray or fugitive gas from deep gas-rich formations has migrated from the subsurface into shallow aquifers and how it has affected groundwater quality. Examples include impacts of uncemented well annuli in areas of historic drilling operations, effects related to poor cement bonding in both new and old hydrocarbon wells, and ineffective cementing practices. We also summarize studies describing how structural features influence the role of natural and induced fractures as contaminant fluid migration pathways. On the basis of these studies, we identify two areas where field-focused research is urgently needed to fill current science gaps related to unconventional gas extraction: (1) baseline geochemical mapping (with time series sampling from a sufficient network of groundwater monitoring wells) and (2) field testing of potential mechanisms and pathways by which hydrocarbon gases, reservoir fluids, and fracturing chemicals might potentially invade and contaminate useable groundwater.

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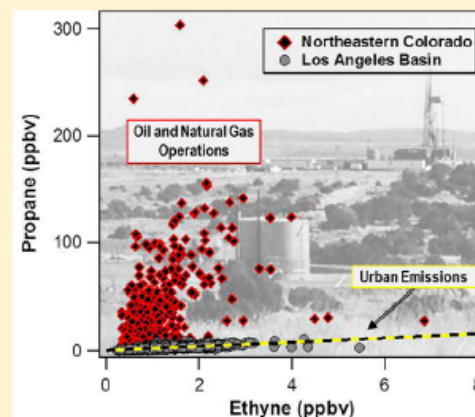
## Source Signature of Volatile Organic Compounds from Oil and Natural Gas Operations in Northeastern Colorado

J. B. Gilman,\* B. M. Lerner, W. C. Kuster, and J. A. de Gouw

Cooperative Institute for Research in Environmental Sciences, University of Colorado, Boulder, Colorado, United States  
NOAA Earth System Research Laboratory, Chemical Sciences Division, Boulder, Colorado, United States

**S** Supporting Information

**ABSTRACT:** An extensive set of volatile organic compounds (VOCs) was measured at the Boulder Atmospheric Observatory (BAO) in winter 2011 in order to investigate the composition and influence of VOC emissions from oil and natural gas (O&NG) operations in northeastern Colorado. BAO is 30 km north of Denver and is in the southwestern section of Wattenberg Field, one of Colorado's most productive O&NG fields. We compare VOC concentrations at BAO to those of other U.S. cities and summertime measurements at two additional sites in northeastern Colorado, as well as the composition of raw natural gas from Wattenberg Field. These comparisons show that (i) the VOC source signature associated with O&NG operations can be clearly differentiated from urban sources dominated by vehicular exhaust, and (ii) VOCs emitted from O&NG operations are evident at all three measurement sites in northeastern Colorado. At BAO, the reactivity of VOCs with the hydroxyl radical (OH) was dominated by C<sub>2</sub>–C<sub>6</sub> alkanes due to their remarkably large abundances (e.g., mean propane = 27.2 ppbv). Through statistical regression analysis, we estimate that on average 55 ± 18% of the VOC–OH reactivity was attributable to emissions from O&NG operations indicating that these emissions are a significant source of ozone precursors.



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## An Evaluation of Water Quality in Private Drinking Water Wells Near Natural Gas Extraction Sites in the Barnett Shale Formation

Brian E. Fontenot,<sup>†,⊥,||</sup> Laura R. Hunt,<sup>†,⊥,||</sup> Zacariah L. Hildenbrand,<sup>†,⊥</sup> Doug D. Carlton Jr.,<sup>†,⊥</sup> Hyppolite Oka,<sup>†</sup> Jayme L. Walton,<sup>†</sup> Dan Hopkins,<sup>‡</sup> Alexandra Osorio,<sup>§</sup> Bryan Bjorndal,<sup>§</sup> Qinhong H. Hu,<sup>†</sup> and Kevin A. Schug<sup>\*,†</sup>

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<sup>‡</sup>Geotech Environmental Equipment Inc., Carrollton, Texas 75006, United States

<sup>§</sup>Assure Controls Inc., Vista, California 92081, United States

**S** Supporting Information

Table 1. Concentrations of Constituents in Barnett Shale Private Water Well Samples<sup>a</sup>

	historical data (1989–99)				active extraction area wells (N = 91)				nonactive and reference area wells (N = 9)			
	N	range	mean ± std error	% ≥ MCL	N	range	mean ± std error	% ≥ MCL	N	range	mean ± std error	% ≥ MCL
TDS	344	129–3302	670.3 ± 21.5	61	91	200–1900	583.1 ± 35.1*	54.9	9	400–600	500 ± 31.6	77.8
arsenic	241	1–10	2.8 ± 0.1	0	90	2.2–161.2	12.6 ± 2.2*	32.2	9	4.7–9.0	6.9 ± 0.7*	0
selenium	329	0.1–50	3.9 ± 0.2	0.3	10	1.0–108.7	33.3 ± 10.5*	20				
strontium	99	20–16700	1028.9 ± 213.7	N/A <sup>b</sup>	90	66.2–18195	2119.8 ± 330.1*	N/A	9	52.4–7646.2	1610 ± 787.1	N/A
barium	337	0.1–382	37.2 ± 2.9	0	90	1.8–173.7	32.3 ± 3.3*	0	9	2.9–60	22.4 ± 11.3*	0
methanol				N/A	24	1.3–329	33.6 ± 13.3	N/A	5	1.2–62.9	27.4 ± 13.7	N/A
ethanol				N/A	8	1–10.6	4.5 ± 1.2	N/A	4	2.3–11.3	6.8 ± 2.4	N/A

<sup>a</sup>All values are measured in  $\mu\text{g/L}$ , except total dissolved solids (TDS), methanol, and ethanol in  $\text{mg/L}$ . Values denoted by asterisks represent statistically significant differences from historical data values (Mann-Whitney U pair wise analysis;  $p < 0.05$ ). Historical data for the counties sampled in this study were obtained online at [www.TWDB.state.TX.us/groundwater/](http://www.TWDB.state.TX.us/groundwater/). Maximum Contaminant Limits (MCL) obtained from the Environmental Protection Agency's (EPA) National Primary Drinking Water Regulations, 2009. TDS MCL = 500  $\text{mg/L}$ , arsenic MCL = 10  $\mu\text{g/L}$ , selenium MCL = 50  $\mu\text{g/L}$ , barium MCL = 2000  $\mu\text{g/L}$ , N/A indicates no MCL has been established. <sup>b</sup>EPA recommends stable strontium values in drinking water do not exceed 4000  $\mu\text{g/L}$ .

accidents such as faulty gas well casings.

# Increased methane in drinking water wells

Robert B. Jackson<sup>a,b,1</sup>, Avner Vengosh<sup>a,b</sup>,  
Stephen G. Osborn<sup>d</sup>, Kaiguang Ren<sup>a</sup>

<sup>a</sup>Division of Earth and Ocean Sciences, <sup>b</sup>Department of Earth and Environmental Science, University of California, Pomona, CA 91768

Edited by Susan E. Trumbore, Max Planck Institute for Chemistry

Horizontal drilling and hydraulic fracturing for natural gas production, but their potential for methane leakage is controversial. We analyzed 141 drinking water wells in the Pennsylvania Plateaus physiographic province to determine the relationship between methane concentration and proximity to shale gas wells. Methane concentrations in drinking water samples, with the exception of one well, were higher for homes <1 km from shale gas wells (mean methane was 23 times higher in wells <1 km from a gas well than in wells >1 km from a gas well,  $P = 0.0013$ ); propane was detected in approximately 1 km distance ( $P = 0.0013$ ); ethane was proposed to influence gas concentrations (distances to gas wells, valley floor elevation, and tectonic Front, a proxy for tectonic activity) was highly significant for methane concentration (multiple regression), whereas distance to the nearest gas well was not significant.

WWW.PNAS.ORG

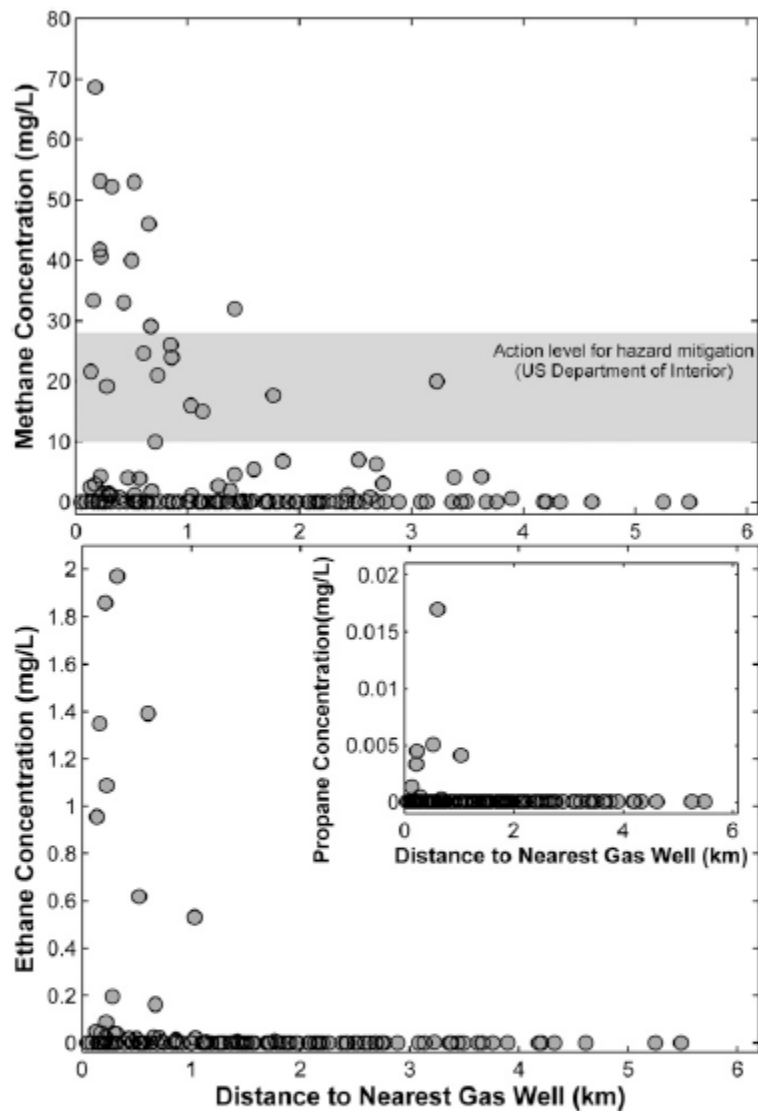


Fig. 1. Concentrations of (Upper) methane, (Lower) ethane, and (Lower Inset) propane (milligrams liter<sup>-1</sup>) in drinking water wells vs. distance to natural gas wells (kilometers). The locations of natural gas wells were obtained from the Pennsylvania DEP and Pennsylvania Spatial Data Access databases (54). The gray band in Upper is the range for considering hazard mitigation recommended by the US Department of the Interior (10–28 mg CH<sub>4</sub>/L); the department recommends immediate remediation for any value >28 mg CH<sub>4</sub>/L.

# of drinking water wells

Robert B. Jackson<sup>a,b</sup>, Robert J. Poreda<sup>c</sup>,

<sup>a</sup>Department of Earth and Environmental Science, <sup>b</sup>Department of Earth and Environmental Science, University of California, Pomona, CA 91768; <sup>c</sup>Department of Earth and Environmental Science, California State Polytechnic University, Pomona, CA 91768

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110



## Shale-Gas Plans Threaten China's Water Resources

THE IMPACT OF SHALE-GAS DEVELOPMENT ON American water quality has received wide attention ("Impact of shale gas development on regional water quality," R. D. Vidic *et al.*, Review, 17 May, p. 826), but potential impacts of China's accelerating shale-gas exploration on the nation's water crisis have been largely ignored.

China has the world's largest shale-gas reserves, at 36 trillion m<sup>3</sup> (1). The country has an ambitious plan to produce 6.5 billion m<sup>3</sup> of shale gas by 2015 (2). Thirteen provinces have been selected as priority areas. However, seven of these provinces are already plagued by water shortages, with less than 2000 m<sup>3</sup> available per person, less than one-quarter of the world average. Four of the thirteen provinces are in Southwest China, and two of these have recently experienced severe half-year droughts (3). Shale-gas extraction will compete for limited water resources with agricultural, industrial, and domestic sectors. Hydraulic fracturing (fracking), the most widely used extraction method in China, consumes large volumes of water mixed with a range of additives. Due to complex geological conditions, Chinese shale-gas wells each consume 10,000 to 24,000 m<sup>3</sup> of water (4, 5). The target gas production of 1.5 billion m<sup>3</sup> in Sichuan will require 171 million m<sup>3</sup> of water, equal to 10.5% of the province's domestic water demand (6).

Some 10 to 90% of fracking fluids are returned to the surface (7). Inadequate treatment introduces heavy metals, acids, pesticides, and other hazardous materials to soil and aquatic environments (8). This will exacerbate China's polluted water environment (9, 10).

Exploitation of China's shale-gas reserves offers opportunities to satisfy the nation's growing energy demands and reduce carbon emissions, but careful management and legislation will be required to avoid shortages and pollution of already stretched water resources.

HONG YANG,<sup>1\*</sup> ROGER J. FLOWER,<sup>2</sup>  
JULIAN R. THOMPSON<sup>2</sup>

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## China has the world's largest shale-gas

Shale-gas extraction will compete for limited

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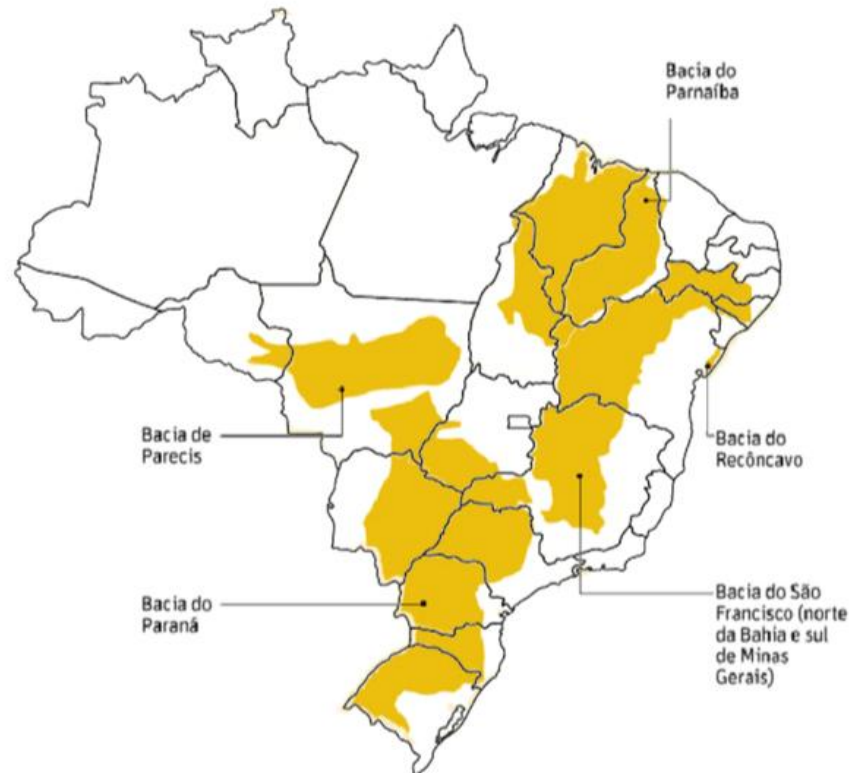
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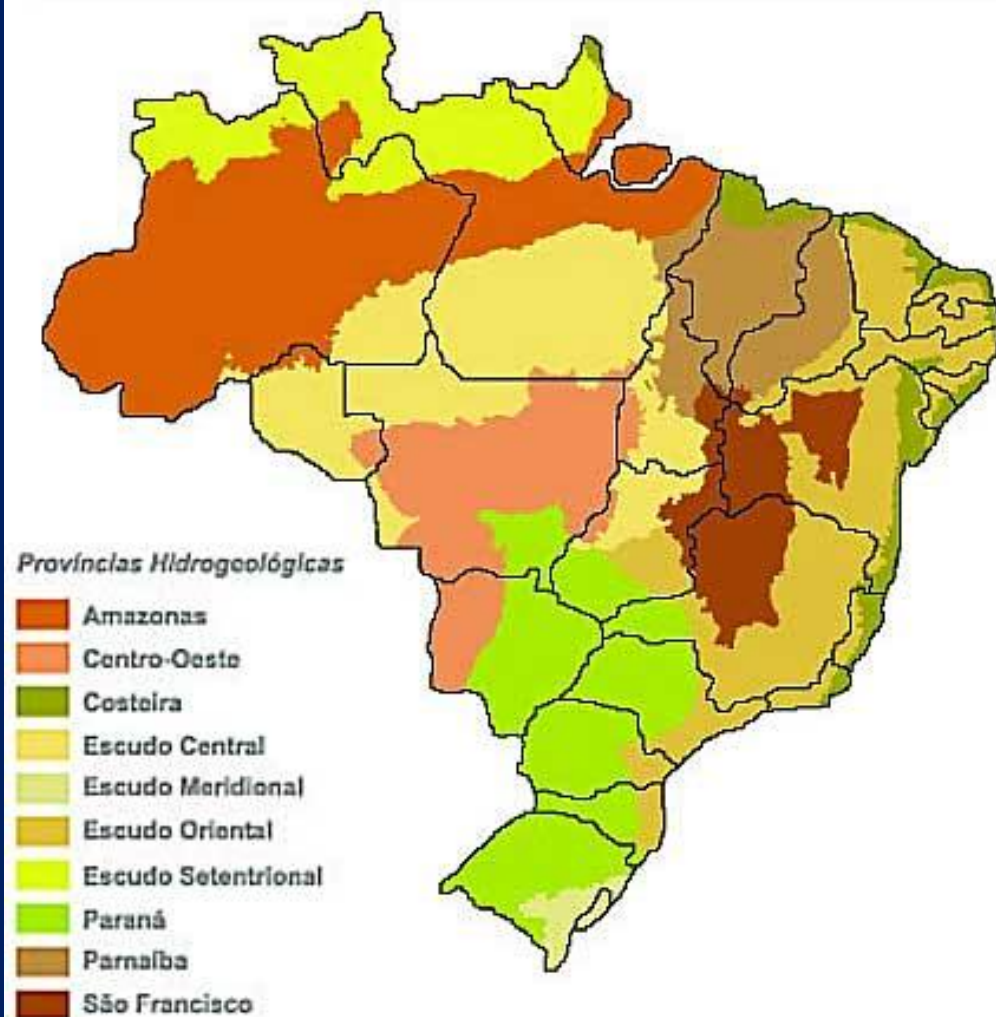
JULIAN R. THOMPSON<sup>2</sup>



## BACIAS SEDIMENTARES COM POTENCIAL DE GÁS NÃO CONVENCIONAL NO BRASIL



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## REPRESENTAÇÃO ESQUEMÁTICA DAS PROVÍNCIAS HIDROGEOLOGICAS DO BRASIL

FONTE: Adaptado de DNPM/CPRM, 1993, citado em MMA, 2003



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## SBPC e ABC enviam carta à presidente Dilma Rousseff solicitando a suspensão da licitação para a exploração do gás de xisto

A Sociedade Brasileira para o Progresso da Ciência (SBPC) e a Academia Brasileira de Ciências (ABC) enviaram, hoje, carta à presidente da República, Dilma Rousseff, manifestando a sua preocupação com o anúncio da Agência Nacional do Petróleo (ANP) da decisão de incluir o chamado “Gás de Xisto”, obtido por fraturamento da rocha (*shale gas fracking*), na próxima licitação, em novembro, de campos de gás natural em bacias sedimentares brasileiras.

No documento, a presidente da SBPC, Helena Nader, e o presidente da ABC, Jacob Palis, justificam sua preocupação pelo fato de que a exploração econômica do gás de xisto vir sendo muito questionada pelos riscos e danos ambientais envolvidos. Por isso, eles solicitam que a presidente suste a licitação de áreas para exploração de gás de xisto, na 12ª Rodada prevista para novembro próximo, por um período suficiente para aprofundar os estudos, realizados por universidades e institutos de pesquisa públicos, sobre a real potencialidade da utilização do método da fratura hidráulica para a retirada do produto das rochas e os possíveis prejuízos ambientais causados por isso.

Também foi enviada cópia da carta para os presidentes da Câmara e do Senado, a ANP, o Centro Nacional de Pesquisa em Energia e Materiais (CNPEM), o Ministério de Minas e Energia (MME), o Ministério da Ciência, Tecnologia e Inovação (MCTI), o Ministério do Meio Ambiente, o CTPetro, a Financiadora de Estudos e Projetos (FINEP), o Conselho Nacional de Desenvolvimento Científico e Tecnológico (CNPq) e as Sociedades Associadas à SBPC.

Leia abaixo, a íntegra da carta:

São Paulo, 5 de agosto de 2013

SBPC-081/Dir.

65ª Reunião Anual da  
SBPC



Agenda



Código Florestal





**Grato pela  
Atenção !**